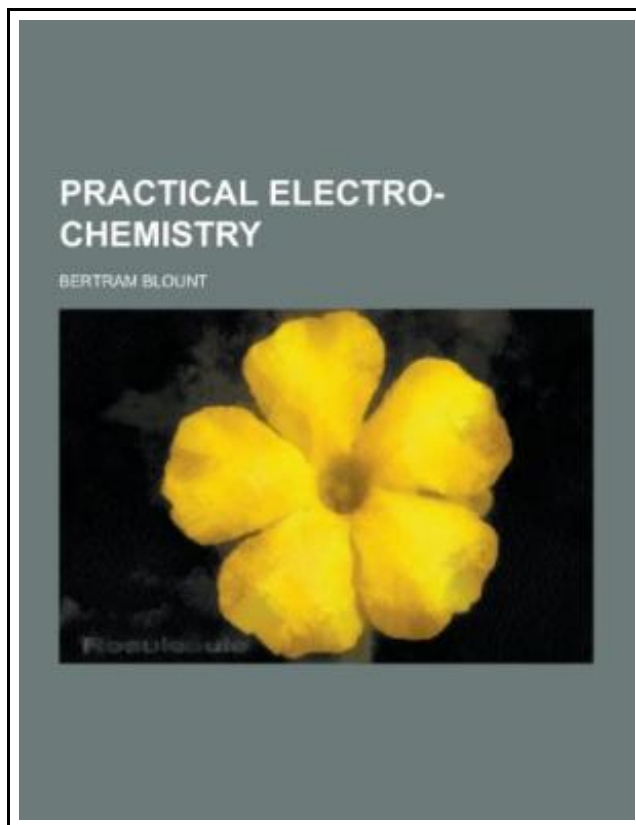


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Theclassics.Us, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1901 edition. Excerpt: . SECTION II Winning and Refining of Metals by Electrolytic Means in Aqueous Solution COPPER THE ELECTROLYTIC REFINING OF COPPER THIS is the largest of all electrolytic industries. It is practised in this country and on the Continent, but the place of its greatest development is the United States; the output there is said to be at least ten times that of all European factories put together. The electrolytic winning of copper, as distinct from its refining, is even now scarcely beyond the experimental stage; an account of the most hopeful processes will be found in a separate section. PEINCIPLES OF THE IELECTROLYTIC REFINING OF COPPER The copper to be refined is already as metal, although the metal is crude, containing not more than 98 per cent. of Cu, and sometimes a smaller percentage. The rationale of electrolytic refining is to transfer this copper, by the selective action of the current, from the anode to the cathode and to leave the impurities behind as a sludge or dissolved in the electrolyte--preferably in the sludge. On first principles it is evident that this mere transference of copper should require no expenditure of energy, because metallic copper is both the raw material and the product; the energy needed to precipitate it from its solution is precisely balanced by the energy set free by its dissolution. This theoretical deduction is entirely consonant with experiment, but there are limitations--commercial and industrial rather than technical--which prevent the full advantage...



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